

MÖSSBAUER AND X-RAY STUDY OF SYNTHETIC LANDAUITE ($\text{Fe}_2\text{Ti}_2\text{O}_7$)

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The oxidation mechanism of ilmenite FeTiO_3 causes an essential interest as for understanding mineral evolution processes as well for producing new inorganic materials. The experimental modeling at the controlled T , P , f_{O_2} – conditions allows one to reconstruct the mineral formation and its successive transformations.

In this work the mechanisms of oxidation at $T = 400 - 950^\circ \text{C}$ in air atmosphere were studied by ^{57}Fe Mössbauer spectroscopy (MS) and X-ray powder diffraction for the sample of ilmenite. The sample were synthesized at $T = 1150^\circ \text{C}$, which is characteristic of the igneous formation processes.

In the temperature interval $T = 600-700^\circ \text{C}$ the ilmenite structure is completely destroyed. At $T = 700^\circ \text{C}$ the landauite ($\text{Fe}_2\text{Ti}_2\text{O}_7$) [1] is formed in the system besides hematite and rutile.

The isomer shift of landauite is characterized for Fe^{3+} ions in the octahedral coordination (0.35-0.36 mm/s relative to metal iron).

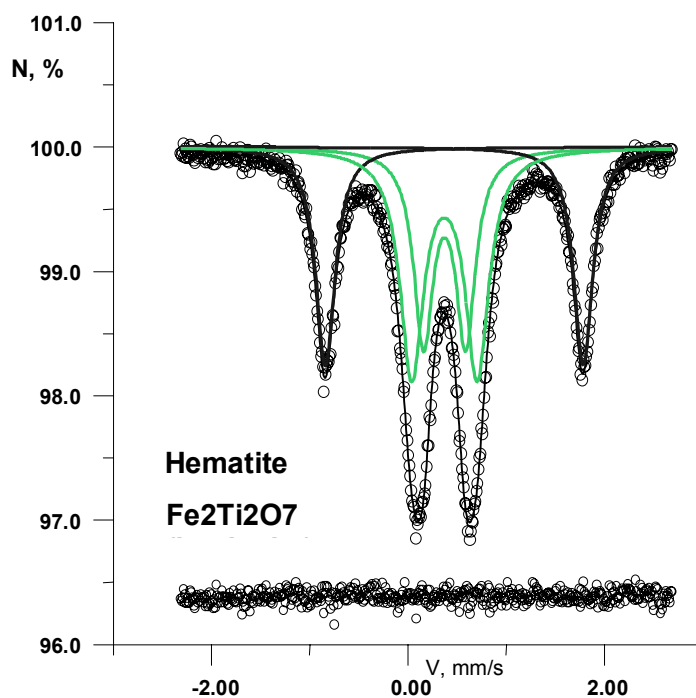


Fig. 1. Mössbauer spectrum of landauite (internal doublet) and hematite (3, 4 lines)

Table 1

X-ray parameters of landauite

[Zn _{0.39} Mn _{0.15} Fe _{0.38}][Ti _{2.87} Fe _{0.13}]O ₇ , [1]			Fe ₂ Ti ₂ O ₇	
<i>d</i> , Å	<i>I/I₀</i>	<i>hkl</i>	<i>d</i> , Å	<i>Hkl</i>
3.36	60	021	3.39	021
3.13	20	220	3.08	220
3.02	60	22 $\bar{1}$	-	-
2.83	100	130	2.88	130
2.72	40	33 $\bar{1}$	-	-
2.59	40	13 $\bar{1}$	-	-
2.45	60	221	2.42	221
2.21	70	311,13 $\bar{2}$	-	-
2.11	90	330	2.12	330
1.946	10	202,13 $\bar{2}$ +	-	-
1.78	80	33 $\bar{2}$, 4 24+	-	-
1.747	10	510,150+	-	-
1.688	50	15 $\bar{1}$	-	-
1.647	10	24 $\bar{2}$, 31 $\bar{3}$ +	1.646	24 $\bar{2}$, 31 $\bar{3}$ +
1.582	80	44 $\bar{1}$	-	-
1.535	10	11 $\bar{3}$	-	-
1.498	50	133	-	-
1.456	10	33 $\bar{1}$, 621+	-	-
1.429	80	-	1.433	
1.244	10	-	-	-

Reference

1. Portnov A.M., Nikolaeva L.E., Stolyarova T.N. // Dokl. Acad. Sci. SSSR. 1966. V.166. N 6. P.1420.